

CYCLIZATION REACTION THROUGH OXIRANE RING CLEAVAGE

OF 9,10-EPOXYHUMULA-2,6-DIEN-13-AL.

SELECTIVE FORMATION OF 3,6-SECOPROTOILLUDANE AND BICYCLOHUMULANE SKELETONS

Haruhisa SHIRAHAMA,* Kiyoharu HAYANO, Gurdial Singh ARORA, Toshikazu OHTSUKA,
Yoshihiko MURATA, and Takeshi MATSUMOTO*

Department of Chemistry, Faculty of Science, Hokkaido University, Sapporo, 060

9,10-Epoxyhumula-2,6-dien-13-al gave 3,6-secoprotioilludane derivatives by treatment with TMSOTf and a bicyclohumulane derivative by $\text{BF}_3 \cdot \text{OEt}_2$ selectively in each case.

Recently biosynthetically patterned selective conversions of humulene 9,10-epoxide (1) to africanol (3) and bicyclohumulenone (4) were performed by trimethylsilyl triflate (TMSOTf) and $\text{BF}_3 \cdot \text{OEt}_2$ respectively by us¹⁾ (Scheme 1). In these conversions, the reactions were initiated by attack of 6,7-double bond to C(9) cationic center to furnish cyclopropane derivatives. Among increasing number of cyclohumulanoids, gem-dimethylcyclopentane derivatives, such as illudins (5), coriolins (6) and pentalenolactones (7), which are constructed through C-C bond formation between C(2) and C(9) of humulene, are more common²⁾ and physiologically active, however. Formation of the C(2)-C(9) bond induced by cleavage of oxirane ring of humulene 9,10-epoxide was therefore attempted (path a of Scheme 2).

In order to achieve attack of 2,3-double bond to the C(9) cationic center, prevention of the attack of 6,7-double bond should be effective. 9,10-Epoxyhumula-2,6-dien-13-al (2) seemed to be suitable for this purpose.

The aldehyde 2³⁾ (mp 126-127°) was prepared from the epoxide 1 by $\text{SeO}_2 / \text{tBuO}_2\text{H}$ (45%)⁴⁾. Geometry of the 6,7-double bond of 2 was confirmed as Z by reduction of 2 to the original epoxide 1 (1. LiAlH_4 , 2. $\text{MsCl}/\text{Et}_3\text{N}/\text{CH}_2\text{Cl}_2$, 3. LiAlH_4). The epoxyaldehyde 2 was then treated with TMSOTf⁵⁾ in toluene at -20° for 2 h and the resulting mixture was desilylated with 0.1N-HCl/MeOH. The product was chromatographed on silica gel column to give two crystalline

hydroxy-aldehydes $\underline{13}^{3)}$ (mp 119-120°; 31%) and $\underline{14}^{3)}$ (mp 90-91°; 47%). The structure of $\underline{13}$ was revealed by coincidence of its acetate $\underline{15}$ with the compound derived from a known secoprotoilludane $\underline{17}^{6)}$. The other hydroxy-aldehyde $\underline{14}$ was also assigned to the hydroxysecoprotoilludadienal structure, since $\underline{14}$ showed in its nmr spectrum a doublet at δ 3.22 ($J=8.5$) due to a proton α to a hydroxyl group and the corresponding ketone $\underline{16}^{3)}$ obtained by Collins oxidation of $\underline{14}$ exhibited an ir absorption at 1745 cm^{-1} due to a five membered ketone. E-Geometry of the 6,7-double bond of both aldehydes $\underline{13}$, $\underline{14}$ was uncovered in their nmr spectra by a triplet peak ($J=5\sim 7$ Hz) due to an olefinic proton β to aldehyde group, since it was known that an trans-cyclooctene showed a double doublet peak ($J=5, 11$ Hz) due to an olefinic proton, while cis-isomer exhibited a triplet peak⁷⁾. The geometry was probably changed through a species $\underline{18}$. The cyclization was thus successfully achieved as expected and gave synthetically useful functionalized 3,6-secoprotoilludane derivatives selectively.

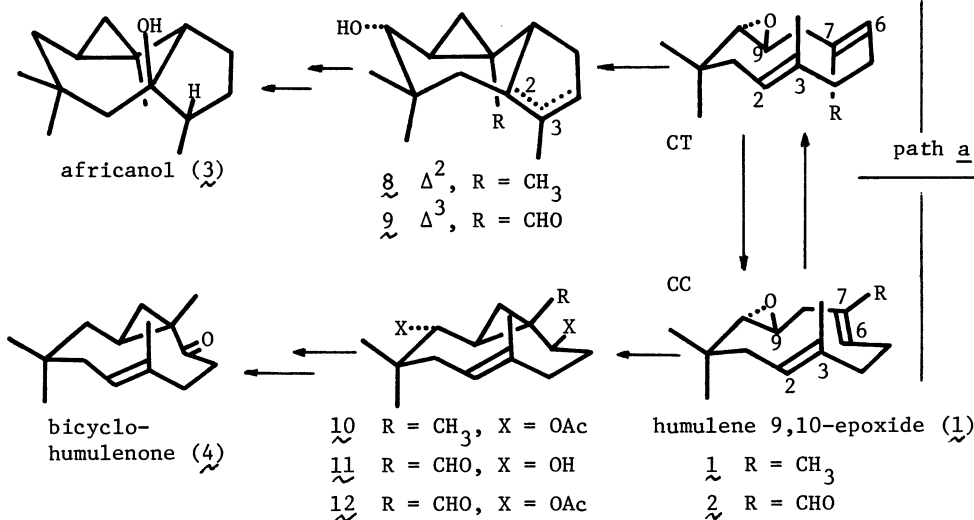
Contrary to the above results the aldehyde $\underline{2}$ was converted to a mixture of two cyclopropane derivatives $\underline{11}^{3)}$ (mp 128-130°, 42%) and $\underline{9}^{3)}$ (oil, 15%) on treatment with $\text{BF}_3 \cdot \text{OEt}_2$ in a mixture of Ac_2O and AcOH (5:1)⁸⁾ at -40° for 40 min and subsequent refluxing with Na_2CO_3 in $\text{MeOH-H}_2\text{O}$ for 12 h. Structures of $\underline{11}$ and $\underline{9}$ were revealed by careful analysis of their nmr spectra and configurations were confirmed by close similarity of the spectra of $\underline{12}^{3)}$ (acetate of $\underline{11}$) and $\underline{9}$ to those of the known compounds $\underline{10}$ and $\underline{8}$ respectively. A crude product of the BF_3 -cleavage reaction showed nmr singlet signals at δ 6.7 and 6.8 before alkaline hydrolysis. This implied initial formation of the diacetylhydrate form ($-\text{CH}(\text{OAc})_2$) of the aldehyde $\underline{2}$ and its subsequent cyclization. Therefore cyclopropane formation took place selectively in the same manner as observed in the previous reaction¹⁾ of $\underline{1}$.

The reactions described here, are a further, but still rare example of reagent-dependent, regio- and stereoselective cyclization of humulene to skeletally different cyclohumulanoids.

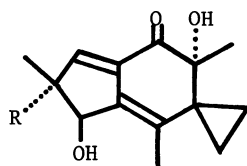
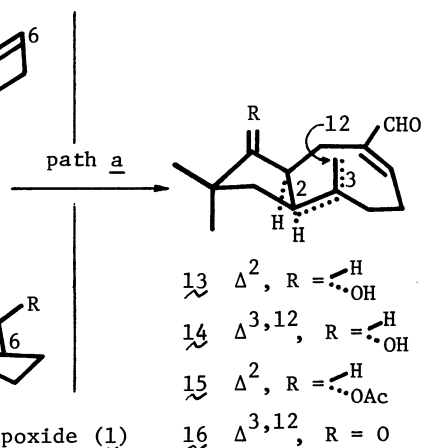
References

- 1) H. Shirahama, K. Hayano, Y. Kanemoto, S. Misumi, T. Ohtsuka, N. Hashiba, K. Furusaki, S. Murata, R. Noyori, T. Matsumoto, *Tetrahedron Lett.*, 21, 4835 (1980).

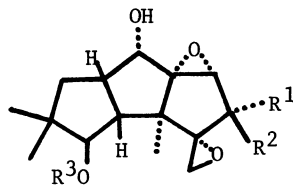
Scheme 1



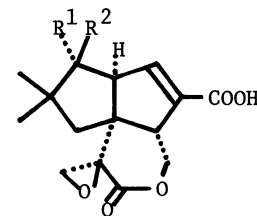
Scheme 2



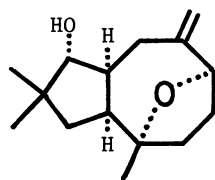
illudin (5)
 M R = CH₃
 S R = CH₂OH



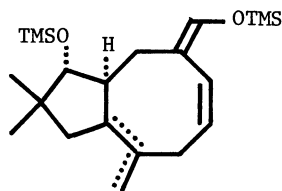
coriolin (6)
 A R¹ = R² = O, R³ = H
 B R¹ = OH, R² = H, R³ = COC₇H₁₅
 C R¹ = R² = O, R³ = COCHOHC₆H₁₃



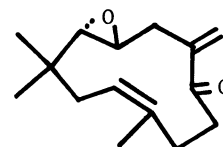
pentalenolactone (7)
 G R¹ = R² = O
 H R¹ = OH, R² = H



$\underline{17}$



$\underline{18}$



$\underline{19}$

2) See the footnote 5 of ref. 1.

3) Satisfactory spectral data were obtained for all new compounds.

Nmr data are shown below. Unless otherwise stated, the nmr spectra were obtained on a 60 MHz instrument.

- 2: δ 0.71, 1.07 (each 3H, s), 1.54 (3H, bs), 5.0 (1H, dd, $J=4$, 9.5 Hz), 6.43 (1H, t, $J=8.5$ Hz), 10.1 (1H, s).
- 9: (400 MHz) δ 0.92 (1H, t, $J=13$ Hz), 0.95, 1.02 (each 3H, s), 1.07 (1H, dd, $J=5$, 7 Hz), 1.35 (1H, td, $J=6$, 9 Hz), 1.44 (1H, btd, $J=10$, 13 Hz), 1.63 (1H, t, $J=7$ Hz), 1.64 (3H, bs), 1.79 (1H, dd, $J=3$, 13 Hz), 2.23 (1H, bdd, $J=10$, 16 Hz), 2.78 (2H, bm), 3.32 (1H, d, $J=9$ Hz), 5.38 (1H, bs), 9.02 (1H, d, $J=1.5$ Hz).
- 11: (400 MHz) δ 1.06 (3H, s), 1.07 (3H, s), 1.34 (1H, dd, $J=5.9$ Hz), 1.52 (3H, bs), 1.65 (1H, dd, $J=5$, 7 Hz), 1.79 (1H, ddd, $J=5$, 8, 9 Hz), 1.83 (1H, bd, $J=14$ Hz), 1.93 (1H, tdd, $J=3$, 6, 14 Hz), 2.12 (1H, dt, $J=3$, 14 Hz), 2.21 (1H, dd, $J=12$, 14 Hz), 2.33 (1H, bd, $J=14$ Hz), 2.66 (1H, ddt, $J=3$, 11, 14 Hz), 2.87 (1H, bd, $J=11$ Hz), 3.36 (1H, d, $J=5$ Hz), 5.33 (1H, bd, $J=12$ Hz), 9.48 (1H, d, $J=1.5$ Hz).
- 12: (100 MHz) δ 0.92, 1.14, 1.56, 2.00, 2.05 (each 3H, s), 4.20 (1H, bd, $J=10$ Hz), 4.9 (1H, d, $J=8$ Hz), 5.37 (1H, bd, $J=12$ Hz), 9.36 (1H, s).
- 13: δ 0.9, 1.0 (each 3H, s), 1.58 (3H, bs), 6.66 (1H, t, $J=5$ Hz), 9.4 (1H, s).
- 14: δ 0.94, 1.10 (each 3H, s), 3.22 (1H, d, $J=8.5$ Hz), 4.85, 5.06 (each 1H, bs), 6.77 (1H, t, $J=6.5$ Hz), 9.37 (1H, s).
- 15: δ 0.89, 0.98, 2.15 (each 3H, s), 1.59 (3H, bs), 4.45 (1H, d, $J=10$ Hz), 6.60 (1H, t, $J=5$ Hz), 9.37 (1H, s).
- 16: δ 1.12, 1.16 (each 3H, s), 4.86, 4.94 (each 1H, s), 6.65 (1H, t, $J=7$ Hz), 9.30 (1H, s).
- 19: δ 0.69, 1.02 (each 3H, s), 1.61 (3H, bs), 5.0 (1H, dd, $J=3$, 9 Hz), 5.39, 5.78 (each 1H, s).
- 4) Besides 15% of ketone 19³⁾ was obtained. Oxidation with SeO_2 -silica gel/ tBuO_2H gave 9,10-epoxyhumula-2,6-dien-13-ol and -12, 13-diol in 4:6 ratio. See B. R. Chhabra, K. Hayano, T. Ohtsuka, H. Shirahama, T. Matsumoto, Chem. Lett., 1981, 1703.
- 5) R. Noyori, S. Murata, M. Suzuki, Tetrahedron, 23, 3899 (1981).
- 6) K. Sakai, T. Ohtsuka, S. Misumi, H. Shirahama, T. Matsumoto, Chem. Lett., 1981, 355. The conversion of 17 to 15 was carried out by the following series of reactions: 1. Me_3SiCl /imidazole, 2. $\text{Li}/\text{EtNH}_2/\text{THF}$, 3. $\text{CH}_3\text{I}/\text{NaH}$, 4. $1\text{N-HCl}/\text{MeOH}$, 5. $\text{Ac}_2\text{O}/\text{Pyr}$, 6. $\text{tBuO}_2\text{H}/\text{SeO}_2/\text{CH}_2\text{Cl}_2$, 7. HCO_2H . The procedure will be described in a full paper.
- 7) K. Hayano, Y. Ohfuné, H. Shirahama, T. Matsumoto, Helv. Chim. Acta, 64, 1347 (1981).
- 8) The reaction employing Ac_2O singly as a solvent gave a complex mixture from which 18% of 11 was solely isolated.

(Received June 21, 1982)